

1 **Immobilised lipase for *in vitro* lipolysis experiments**

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18 **Keywords:** immobilised lipase, lipid digestion, *in vitro* lipolysis, lipid-based drug delivery, synchrotron
19 small angle X-ray scattering

20

21 **Abbreviations:**

22 BS, bile salt

23 CALB, lipase B from *Candida antarctica*

24 FA, fatty acid

25 SAXS, small angle X-ray scattering

26 TBU, tributyrin units

27 **Abstract**

28 *In vitro* lipolysis experiments are used to assess digestion of lipid-based formulations, and probe
29 solubilisation by colloidal phases during digestion. However proteins and other biological
30 components in the pancreatin often used as the lipase result in high background scattering when
31 interrogating structures using scattering approaches, complicating the resolution of colloidal
32 structures. In this study, to circumvent this problem, a modified *in vitro* digestion model employing
33 lipase immobilized on polymer beads, which allows for separation of the lipid digestion components
34 during lipolysis, was investigated. Titration of the fatty acids released during digestion of medium
35 chain triglycerides using pancreatin compared to immobilized lipase, combined with HPLC was used
36 to follow the digestion, and small angle X-ray scattering was used to determine colloidal structure
37 formation. Digestion of medium chain triglycerides at the same nominal activity revealed that for the
38 immobilized lipase, a longer digestion time was required to achieve the same extent of digestion.
39 However, the same structural endpoint was observed, indicating that structure formation was not
40 affected by the choice of lipase used. Lipolysis with immobilized lipase led to the reduction of
41 parasitic scattering, resulting in clearer and more defined scattering from the structures generated
42 by the lipolysis products.

43

44 **Introduction**

45 Lipid based formulations (LBF) are of interest for the delivery of poorly water soluble drugs. During
46 the lipolysis of triglycerides, monoglycerides (MG) and fatty acids (FA) are produced. These combine
47 with endogenous amphiphilic molecules in the gastrointestinal tract to form bile salt/phospholipid
48 mixed micelles and other liquid crystalline phases. These structures are important in maintaining
49 drug solubilisation in the gastrointestinal environment, leading to enhanced absorption and
50 bioavailability of co-administered drugs [1, 2].

51

52 *In vitro* lipolysis experiments using a pH stat apparatus are frequently utilised to evaluate digestion
53 of LBF to examine drug disposition, solubilisation and precipitation, with the aim to achieve *in vitro*-
54 *in vivo* correlations. The study of lipolysis from a compositional aspect remains largely reliant on the
55 titration profile. The pH stat follows the progress of triglyceride digestion, by titrating against the
56 fatty acids liberated.

57

58 Understanding structure formation during lipolysis remains a major hurdle, due to lack of real time
59 methods to elucidate structural changes during the dynamic process, and this has recently been
60 reviewed [3]. Until recently, studies have been limited to microscopy, scattering and spectroscopic

61 techniques to investigate equilibrium structures that are assembled from the individual components
62 representing the end point of lipid digestion. The field has since progressed to the use of *in vitro*
63 lipolysis models coupled to synchrotron small angle X-ray scattering [4, 5], where the digest medium
64 is flowed through a capillary placed in the X-ray beam for probing of nanostructure formation in the
65 millisecond time frame.

66
67 The current protocol for *in vitro* digestion studies often utilizes porcine pancreatin extract [4, 6-8] as
68 the lipase source to simulate digestion in the small intestine. However there are many proteins and
69 enzymes in the crude extract which contribute to high undesired scattering from pancreatic
70 components, when using techniques such as small angle X-ray scattering (SAXS) and dynamic light
71 scattering (DLS) to understand the structural evolution caused by lipolysis products in these systems.
72 These may obscure scattering from colloidal species produced on digestion, and complicate the
73 resolution of structures that are present.

74
75 Novozym® 435 is a commercially available recombinant lipase B from *Candida antarctica* (CALB)
76 manufactured by Novozymes. It is immobilized by physical adsorption to macroporous polyacrylate
77 resin, which has an average particle size of 315-1000 μm , surface area of 130 m^2/g and pore
78 diameter 150 Å [9, 10]. Immobilized lipases are widely used as biocatalysts in the food, detergent,
79 textile, cosmetic and pharmaceutical industries, for example, hydrolysis of triglycerides to produce
80 fatty acids and esterification, transesterification, aminolysis in organic solvents. Immobilized lipases
81 have been successfully manufactured by ionic binding, covalent binding, cross-linking, entrapment
82 and encapsulation. Advantages of immobilized lipase over free lipase are that they are cost effective
83 and reusable, for continuous large-scale commercial processes, has enhanced chemical and thermal
84 stability and exhibit high enantioselectivity [11]. In the context of *in vitro* lipolysis experiments,
85 having enzyme immobilized on beads enables separation from the digesting medium which may
86 improve the ability to study structural aspects using scattering methods.

87
88 Consequently the aim of this study was to investigate the possibility of using lipase immobilised to
89 polymer beads as an alternative to powdered pancreatin to improve the current method for
90 studying structures during lipolysis. To the best of our knowledge immobilized lipase has not been
91 applied to *in vitro* lipolysis experiments. Digestion of medium chain triglycerides (tricaprylin and
92 Captex 355, a commonly used formulation lipid consisting of a mixture of tricaprylin and tricaprln,
93 were assessed to compare the activity, effect of pH, temperature, and buffer type and resulting
94 structural detail after digestion with immobilized lipase compared to pancreatin extract.

95

96 **Experimental**

97 **Materials**

98 Captex 355 (MCT composed of 59% caprylic acid (C₈), 40% capric acid (C₁₀), < 1% lauric acid (C₁₂) as
99 stated in the product information) was obtained from Abitec Corporation (Janesville, WI, USA) and
100 used without further purification. Tributyrin (>98%) was obtained from TCI Co. Ltd (Kawaguchi City,
101 Saitama, Japan). Tricaprylin (> 99%), Tris maleate (reagent grade), bile salt (sodium
102 taurodeoxycholate, > 95%) and 4-bromophenylboronic acid (4-BPBA, > 95%) were purchased from
103 Sigma Aldrich (St. Louis, MO, USA). Sodium azide was purchased from Merck Schuchardt OHG
104 (Eduard-Buchner-StraÙe, Hohenbrunn, Germany). Phospholipid (1,2-dioleoyl-sn-glycero-3-
105 phosphocholine, DOPC) was from Trapeze Association Pty Ltd (Clayton, Victoria, Australia). USP
106 grade pancreatin extract was purchased from Southern Biologicals (Nunawading, Victoria, Australia).
107 Novozym® 435 was obtained from Novozymes (Bagsvaerd, Denmark). Calcium chloride (> 99%) was
108 obtained from Ajax Finechem (Seven Hills, NSW, Australia). Sodium chloride (> 99%) was purchased
109 from Chem Supply (Gillman, SA, Australia). HPLC grade methanol (MeOH) was purchased from
110 Merck. Trifluoroacetic acid (TFA) was obtained from Pierce (Rockford, Illinois, USA). Water used was
111 sourced from a Millipore water purification system using a Quantum™ EX Ultrapure Organex
112 cartridge (Millipore, Sydney, Australia).

113

114 **Determination of lipolytic activity**

115 The lipolytic activity was determined using the pH stat method, which was used to construct a
116 calibration curve. Digestion buffer typically was prepared with 50 mM Tris maleate, 5 mM
117 CaCl₂·2H₂O, 150 mM NaCl, 6 mM NaN₃ as anti-microbial agent, and adjusted to pH 6.5. The substrate
118 used was 6 g tributyrin [6, 12] dispersed in 10 mL of digestion buffer for immobilized lipase or 9 mL
119 digestion buffer for pancreatic lipase respectively, which itself was dispersed in 1 mL digestion buffer
120 as below. The quantity of digestion buffer used differed depending on the lipase source to maintain
121 a constant buffer:tributyrin ratio.

122

123 Immobilized lipase was added to the test system at different masses; whereas the pancreatic lipase
124 was prepared by weighing the required mass of the pancreatic lipase in a centrifuge tube, adding 5
125 mL digestion buffer to form a suspension, which was adjusted to pH 6.5 by the addition of 30 µL 0.5
126 M-NaOH and 5 mL buffer to form a suspension. The suspension was magnetically stirred for 15 min,
127 and centrifuged for 15 min at 3500 rpm at 25 °C as a compromise to prevent a denaturation of the
128 lipase and a decline in activity, and minimize the temperature difference between the lipase

129 suspension and the digestion medium. Fresh pancreatin extracts were prepared each day and the
130 supernatant was kept at room temperature prior to use.

131

132 The reaction was typically performed at 37 °C and in Tris buffer pH 7.5 and lipase (mg immobilized or
133 1 mL pancreatic lipase suspension, containing 10 000 TBU units) was added to initiate digestion and
134 the reaction was titrated with 0.6 M NaOH. In the case where the influence of buffer type,
135 temperature or pH on activity ~~were~~ was studied, these variables were changed accordingly. The
136 activity was defined in Tributyrin Units (TBU) where 1 TBU is the amount of enzyme used to liberate
137 1 μmol of titratable free fatty acid per minute. A plot of μmol FA titrated per minute yields an initial
138 linear portion, where the slope gives the activity in TBU per mL of digest under the assay conditions.

139

140 ***In vitro* lipolysis**

141 ~~Digestion buffer typically was prepared with 50 mM Tris maleate, 5 mM CaCl₂·2H₂O, 150 mM NaCl, 6~~
142 ~~mM NaN₃ as anti-microbial agent, and adjusted to pH 6.5.~~ A micellar solution was prepared in
143 digestion buffer with bile salt (sodium taurodeoxycholate) and phospholipid (DOPC) at
144 concentrations of 5 mM:1.25 mM ~~to~~ as simulated fasted intestinal fluid [13, 14]. The phospholipid
145 was dissolved in chloroform in a round bottom flask and the chloroform was evaporated off under
146 vacuum to leave a thin film. Bile salt and digestion buffer were added and the solution was placed in
147 a sonicator bath for 30 min, before it was stored at 4 °C to equilibrate overnight. Pancreatin extracts
148 were prepared by adding 4 g of porcine pancreatin powder to 5 mL of digestion buffer to achieve an
149 activity of 10 000 TBU/mL. The suspension was magnetically stirred for 15 min, and centrifuged for
150 15 min at 3500 rpm at 25 °C. ~~Fresh pancreatin extracts were prepared each day and the supernatant~~
151 ~~was kept at room temperature prior to use.~~

152

153 *In vitro* digestion studies were performed using a pH-stat auto titrator (Metrohm, Switzerland),
154 similar to previous reports [4, 6, 8, 15, 16]. Lipid was added to the fasted simulated intestinal fluid in
155 the thermostatted digestion vessel at 37 °C, and magnetically stirred for 5 min for complete mixing
156 and thermal equilibration. The pH was adjusted to 6.5 ± 0.003, chosen as a compromise between the
157 optimum for pancreatic lipase activity pH (6 – 10) [17] and duodenal pH (5.9 – ~~5.5~~) [18, 19]. On
158 addition of lipase (~1000 TBU/mL digest) the pH-stat titrated the digestion with 0.6 M NaOH in order
159 to maintain pH 6.5 ± 0.003. Digestion was allowed to proceed for up to 8 hr, to ensure a similar
160 extent of digestion, in which the degree of enzymatic digestion of the lipid was reflected in the
161 volume of NaOH used to neutralize the fatty acids liberated during the digestion process.

162

163

164 A blank digestion without lipid but with bile salt micelles present was performed as a background
165 experiment, to account for fatty acids that were produced from phospholipids and was subtracted
166 from the profiles for the lipolysis experiments.

167

168 **Lipid composition in digestion medium**

169 A few techniques have been used to separate and analyze lipid digestion products including HPTLC,
170 GC and HPLC-MS. However these have shortcomings such as long run times, labour intensive, high
171 cost, requirement of specialized equipment and use of volatile solvents. Reversed phase HPLC using
172 refractive index detection has recently been shown to be a useful tool to determine lipid
173 composition following *in vivo* digestion that it is simple and non-destructive [20]. Thus this method
174 was employed to separate and quantify fatty acid analytically during digestion for correlation with
175 total fatty acid produced as indicated by the pH stat titration. Samples (200 μ L) were taken during
176 digestion for analysis of fatty acid content by HPLC. To halt digestion in the sample lipase inhibitor
177 (20 μ L of 0.05 M 4-BPBA in methanol) was added to eppendorf tubes, where the methanol was
178 evaporated off prior to addition of samples.

179

180 *Reversed-phase high performance liquid chromatography (HPLC)* - Samples were analysed for fatty
181 acid content and separated by an isocratic reversed-phase HPLC method using a 4.6 \times 150 mm
182 Phenomenex Luna C₈ (2) (5 μ m, 100 Å) analytical column, with a 15 \times 3 mm Brownlee RP-18 (7 μ m)
183 guard column. The HPLC system consisted of a Shimadzu CBM-20A system controller, LC-20AD
184 solvent delivery module, SIL-20A auto sampler and a CTO-20A column oven set at 40°C, coupled to a
185 RID-10A differential refractometric detector (Shimadzu Corp., Kyoto, Japan). An injection volume of
186 40 μ L was used to separate caprylic acid and capric acid using a mobile phase consisting of
187 MeOH/water (75:25 v/v) with 0.1% TFA (v/v of total mobile phase) at a flow rate of 1 mL/min. Assay
188 parameters are shown in Table SI-1.

189

190 *Preparation of standards and samples* - A stock solution of caprylic and capric acid was prepared at a
191 concentration of 10 mg/mL in methanol. A set of standards containing 0.1, 0.2, 0.5, 1.0 and 2.0
192 mg/mL of both lipids was prepared by mixing and dilution of the stock solution in the mobile phase.
193 All stock solutions and standards were stored at 4 °C before analysis. Calibration curves were
194 prepared by plotting the area under the curve against known concentration of standard solutions.
195 Lipolysis samples were diluted in mobile phase prior to HPLC analysis. Unknown sample

196 concentrations were calculated from the standard equation $y = mx + c$, as determined by the linear
197 regression of the unweighted standard curve.

198

199 *Assay validation* - Validation of the HPLC assay was run over three days and the results are shown in
200 Table SI-2. Intra-assay accuracy was determined by replicate analysis (n=5) of standard solutions of
201 lipids at three concentrations (0.1, 1.0 and 2.0 mg/mL). Inter-assay accuracy was determined on
202 three separate days. The data were expressed as a percentage of the measured concentration over
203 the theoretical concentration, where mean accuracy was within $\pm 15\%$ of the theoretical
204 concentration. Intra-assay precision (repeatability) and inter-assay precision (reproducibility) were
205 calculated in all three runs for each lipid at all three concentrations and expressed as the coefficient
206 of variation (% CV) of replicate assays. Linearity was performed on standard curves for each run and
207 linearity was fulfilled when the correlation coefficient (r^2) of the regression line was > 0.99 .

208

209 **Synchrotron small angle X-ray scattering (SAXS)**

210 SAXS measurements were performed at the SAXS/WAXS beamline [21] at the Australian
211 Synchrotron. An X-ray beam with a wavelength of 1.1271 \AA (11 keV) was used. A sample to detector
212 distance of 1588 mm covered the q -range $0.01 < q < 0.7 \text{ \AA}^{-1}$ with q being the magnitude of the
213 scattering vector defined as $q = (4\pi/\lambda)\sin\vartheta/2$, λ being the wavelength and ϑ the scattering angle.

214

215 Samples were drawn into a capillary which was fixed in a thermostatted metal heating block
216 controlled by a Peltier system accurate to $\pm 0.1 \text{ }^\circ\text{C}$, for measurements taken at $37 \text{ }^\circ\text{C}$. The 2D SAXS
217 patterns were acquired over 1 s using a Pilatus 1M detector with active area $169 \times 179 \text{ mm}^2$ and with
218 a pixel size of $172 \text{ }\mu\text{m}$. 2D scattering patterns were integrated into the 1D scattering function $I(q)$
219 using the in-house developed software package scatterBrain. Scattering curves are plotted as a
220 function of relative X-ray intensity, $I(q)$, versus q .

221

222 **Results**

223 **Lipolytic activity**

224 Activity of the immobilized lipase was very different to that of the pancreatin extract (Figure 1). The
225 immobilized lipase displayed consistent lipolytic activity towards tributyrin where it increased
226 linearly with concentration, with approximately 37 TBU/mg immobilized lipase. In contrast, the
227 activity of pancreatic lipase reached a plateau with increasing concentration, approximately 12
228 TBU/mg dry pancreatin powder at saturation.

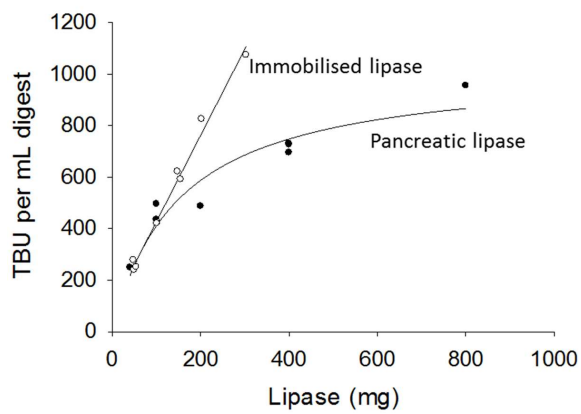
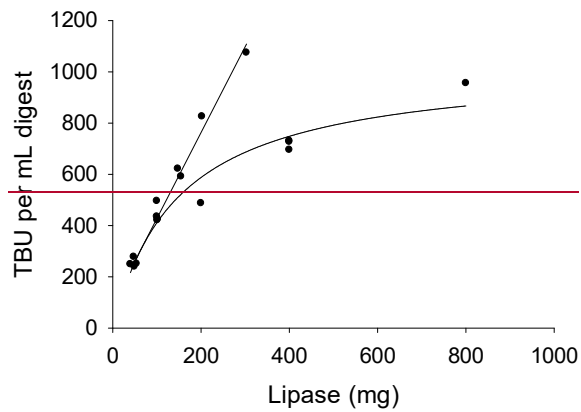
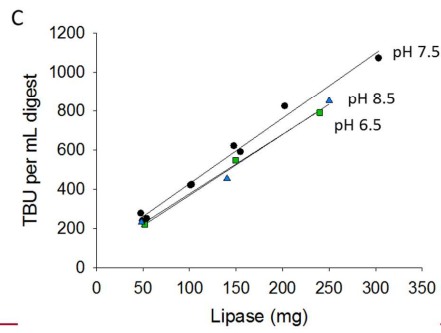
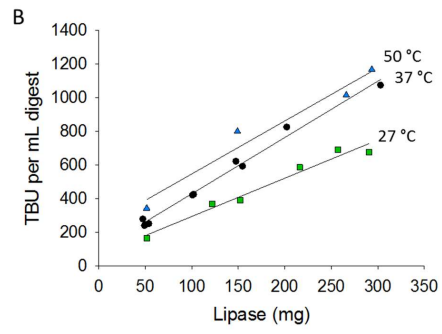
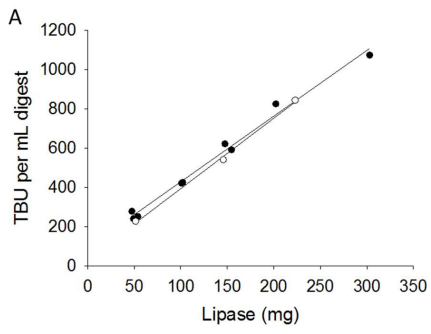


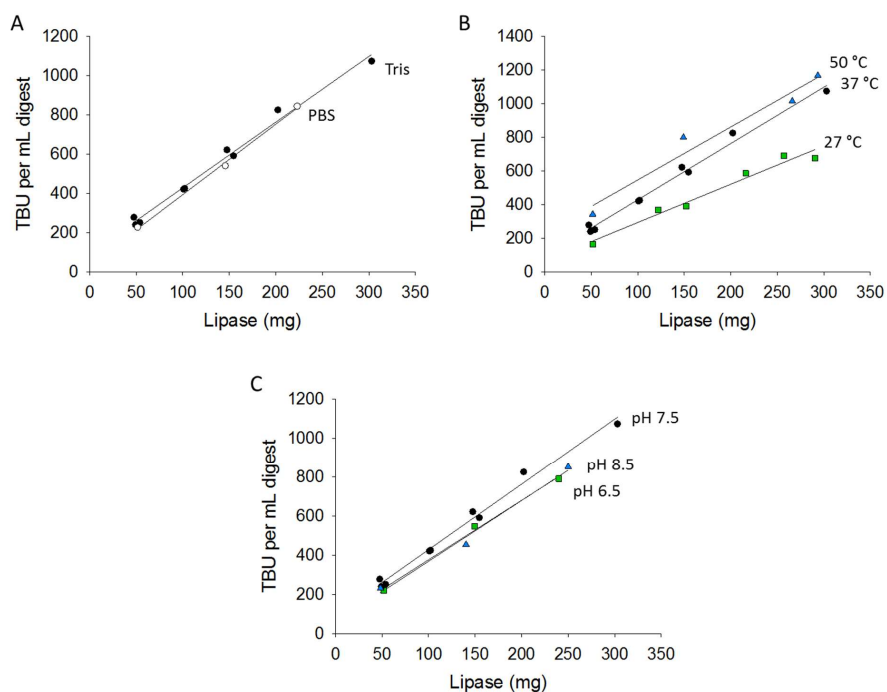
Figure 1: Comparison of activity (TBU) of immobilized (open circles) and pancreatic (closed circles) lipase with increasing mass in Tris buffer pH 7.5 at 37 °C

Factors affecting the activity of immobilized lipase

The influence of buffer, temperature and pH on lipolytic activity of immobilized lipase was assessed to further understand the physico-chemical understanding of digestion, as these factors may affect lipase activity. No significant buffer- or pH-dependency was observed on the activity (Figure 2A and C). Lipase activity was however observed to vary with temperature as reported in the product information (Figure 2B). An increase in activity was observed with increasing temperature as anticipated due to an increase in kinetic energy in the system and lowering of the activation energy required for the reaction, and a decrease in temperature hampered the reaction.

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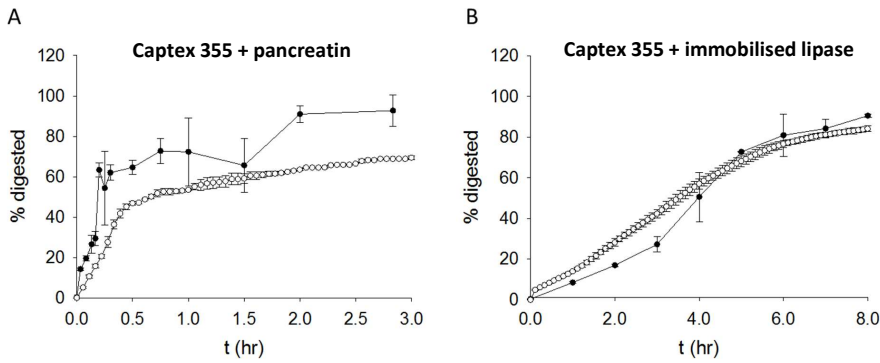
244
 245 **Figure 2:** Effect of reaction variables on activity of immobilized lipase in A) Buffer type (Tris (closed
 246 circles) and PBS (open circles) pH 7.5 at 37 °C). B) Temperature (Tris buffer pH 7.5). C) Buffer pH (Tris
 247 buffer pH 6.5, 7.5 and 8.5 at 37 °C).

248
 249 **Digestion kinetics and quantitation of fatty acid released during *in vitro* lipolysis of medium chain
 250 triglycerides**

251 The kinetics of digestion when the immobilized lipase was added to initiate digestion of tricaprillin
 252 and Captex 355 in the fasted state was slower compared to pancreatic lipase. A longer digestion
 253 time was required to achieve a plateau in the titration curve, indicative of the end of lipolysis of ~50
 254 mM medium chain triglyceride, Captex 355 (Figure 2A3A). The % digested was calculated to be 69.3
 255 and 84.1% for pancreatic and immobilized lipase respectively. This was calculated from the number
 256 of moles of NaOH added during the reaction, on the basis that one mole of triglyceride is hydrolysed
 257 to produce 1 mole of 2-monoglyceride and 2 moles of fatty acid [22, 23]. Notably, the digestion
 258 profiles of Captex 355 for both the lipases had a sigmoidal shape but covered different time scales.

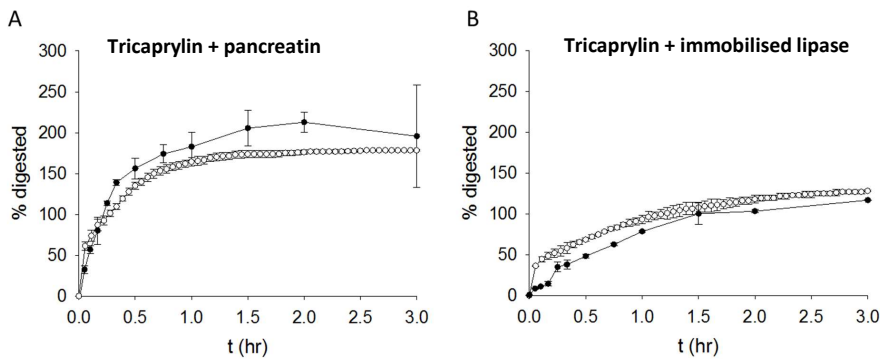
259
 260 A strong correlation between the moles of fatty acid determined by pH stat titration and HPLC with
 261 refractive index detection during lipolysis was observed (Figure 3). However, at the endpoint of

262 digestion, the higher fatty acid content determined by HPLC was higher at 92.7 and 90.6% for
263 pancreatic and immobilized lipase respectively.



264
265 **Figure 3:** Kinetics of digestion, during lipolysis of ~50 mM Captex 355 in the fasted state at 37 °C with
266 A) pancreatic and B) immobilised lipase, showing fatty acid by HPLC (closed circles) and titration
267 (open circles). Data are mean ± range (n=2).
268

269 In the case of lipolysis of 5 mM tricaprylin, digestion proceeded past completion based on the
270 aforementioned assumption of two fatty acids and one monoglyceride (Figure 4). For pancreatic
271 lipase, the extent of digestion was calculated to be 178.4 % by pH stat titration and 195.6% by HPLC
272 compared to 128.2% by pH stat titration and 116.9% by HPLC for the immobilized lipase.



273
274 **Figure 4:** Kinetics of digestion, during lipolysis of 5 mM tricaprylin in the fasted state at 37 °C with A)
275 pancreatic and B) immobilised lipase showing fatty acid by HPLC (closed circles) and pH stat titration
276 (open circles). Data are mean ± range (n=2).
277

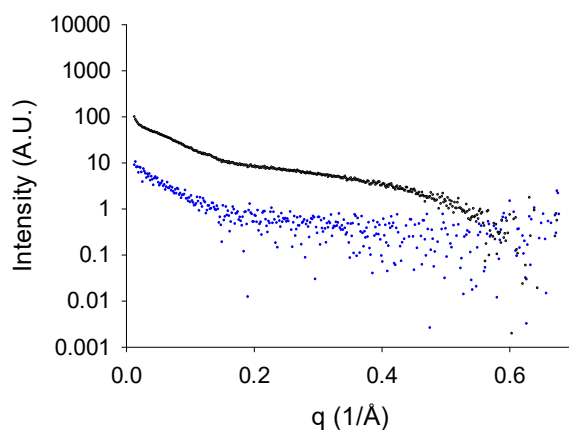
278 To determine whether separation of immobilised lipase from the digestion medium would be
279 sufficient to halt digestion, samples were removed at predetermined timepoints, without addition of

280 the commonly used inhibitor, 4-BPBA. Correlation of fatty acid content by HPLC and pH stat titration
281 again was observed, confirming that the lipase remains adsorbed to the polymer beads rather than
282 leaching into the digestion medium (Figure SI-1).

283

284 **Structure formation during *in vitro* lipolysis**

285 Synchrotron SAXS was used to determine the structures formed at the completion of digestion to
286 test the hypothesis of reduced scattering from the immobilized lipase, and to confirm independence
287 of structures formed. After subtraction of the buffer as the background, a significant order of
288 magnitude reduction in the scattering was observed in the dispersion of immobilized lipase in
289 digestion buffer compared to pancreatic lipase in digestion buffer (Figure 65).

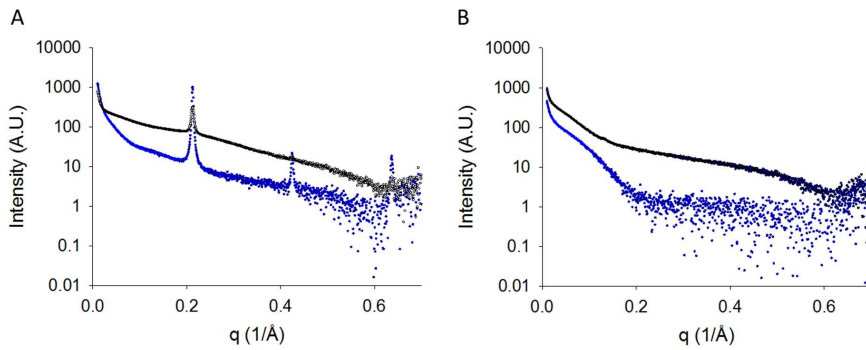


290

291 **Figure 65:** SAXS profiles of pancreatic (black) and immobilized (blue) lipase dispersed in digestion
292 buffer pH 6.5 measured at 37 °C

293

294 The two lipases were then used to initiate lipolysis in *in vitro* digestion of formulation lipids and a
295 sample was removed at the endpoint of digestion for structural characterization by SAXS. After 8 hr
296 lipolysis by immobilized lipase, the peaks indicative of a lamellar phase were observed to be more
297 prominent in the scattering profiles after buffer subtraction compared to the sample digested for 3
298 hr with pancreatic lipase (Figure 7A6A). The lamellar phase had a repeat distance of 29.6 Å. Similarly,
299 better structural resolution of the micellar structure was achieved using immobilised lipase following
300 lipolysis of 5 mM tricapylin under the same digestion conditions (Figure 7B6B).



301
 302 **Figure 76:** SAXS profiles of end point of lipolysis of A) ~50 mM Captex 355 and B) 5 mM tricaprylin in
 303 the fasted state at 37 °C reveal formation of lamellar phase and micelles respectively. Lipolysis was
 304 performed with pancreatic (upper) and immobilized (lower) lipase.

305
 306 **Discussion**

307 Lipases have been employed in the food, dairy, pharmaceutical, detergent, textile, pulp and paper,
 308 animal feed, leather and cosmetics industry. The application of immobilised lipases is increasing due
 309 to advantages over the free form such as enantioselectivity, increased chemical and thermal
 310 stability, cost efficiency and ability for reuse in continuous large-scale commercial processes.
 311 Significant resources have been deployed in recent times to attempt to improve and standardize *in*
 312 *vitro* digestion experiments to assist the development of LBFs [8]. It is therefore surprising that
 313 immobilized lipases have not been investigated in these activities.

314
 315 **Factors affecting lipase activity**

316 The lipolytic activity of the immobilised lipase was characterized to determine the influence of
 317 buffer, pH and temperature. The independence of lipase activity on pH and buffer type opens new
 318 avenues for improved and more versatile versions of the ‘accepted’ *in vitro* digestion models.
 319 Specifically, more intestinally-relevant buffer systems, at pH values closer to those expected in the
 320 duodenum, rather than those traditionally employed in such test methods, would be expected to
 321 not deviate from the enzyme activity and structure formation while adding versatility to the choice
 322 of model variables. The adherence to past protocols may be claimed to have to a degree actually
 323 held back the research field through restricted access to components such as pancreatic lipase and
 324 specific bile components. Challenging these accepted norms through alternative approaches to the
 325 models will help to broaden our understanding of the lipid based formulation field.

326

327 **Immobilised lipase displays differences in digestion kinetics**

328 This is the first description of immobilised lipases used for *in vitro* lipolysis. Formerly, the definition
329 of a “true” lipase included the presence of a lid and activation by the presence of an interface, such
330 that increased activity is observed when the substrate is present as an emulsion [24]. However this
331 definition has been revised to carboxylesterases that catalyse the hydrolysis and synthesis of
332 acylglycerols of chain length of 10 carbon atoms or more [23-25].

333

334 Pancreatic lipase is a typical lipase in that the active site is shielded from solvent, but when it is
335 exposed to a hydrophobic interface, it undergoes a conformational change where the active site is exposed.
336 This is known as resulting in interfacial activation [26-29]. Co-lipase is also required to bind to pancreatic lipase to allow it to act at the oil/water interface to hydrolyse
337 triglycerides, and remove bile acids from the interface which could inhibit further digestion [30]. In
338 the current study, the immobilized lipase used, CALB, has a typical lipase α/β hydrolase fold, and the
339 active site consists of a serine, aspartic acid and histidine catalytic triad and no typical lid domain.
340 The amino acids are the same as those found in human and porcine pancreatic lipase [31]. Despite
341 this difference, the current work confirms the lipolytic activity of CALB, and is in agreement with
342 literature reports of little or no interfacial activation [24, 26]. It was found to display enhanced
343 lipolytic activity against tributyrin compared to pancreatic lipase. This is potentially due to
344 differences in conformation; immobilized lipase is adsorbed onto hydrophobic support, which
345 resembles the substrate, thus rendering the lipase in a fixed “on” conformation [32, 33].

347

348 The difference in extent of digestion by the two lipases for tricaprylin could be due to the specificity
349 of the lipase. Novozym® 435 can show positional specificity, according to the product specifications,
350 whereas the porcine pancreatic extract contains other enzymes such as Pancreatic Lipase-Related
351 Protein 2, Pancreatic Phospholipase A2, co-lipase and cholesterol esterase so it is possible that
352 further digestion occurs as a result. The lipid composition during digestion of Captex 355 with
353 pancreatic lipase is in agreement with quantification of lipid content during medium chain digestion
354 by HPTLC [34]. It has also been previously reported that after digestion, the MG:FA ratio could be
355 higher than 1:2 as high as 1:6 [35-37].

356

357 Captex 355 is a mixture of medium chain triglycerides commonly used in lipid based formulations
358 [38, 39]. Digestion of Captex 355 with immobilised lipase was slower than with pancreatin. This
359 could be attributed to the lack of access of the lipase active site to the substrate due to confinement
360 to the polymer beads compared to the pancreatic lipase dispersed freely in solution. The likely

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361 mechanism of triglyceride digestion by immobilized lipase is at the droplet interface, as the majority
362 of the triglyceride is in the emulsion droplets due to their low solubility in water-by collision of the
363 beads with lipid droplets. If this is the case, it would be reasonable to expect that if the bead size
364 was reduced and the surface area was increased at a nominal enzyme activity, then the rate of
365 digestion with longer chain triglycerides would increase.

366
367 Interestingly, the immobilized lipase was found to be very effective in digesting the short chain lipid,
368 tributyrin, compared to tricaprylin. It has been proposed in literature that lipolysis is dependent on
369 solubility of substrate in the solvent [40]. The difference in activity can be explained by the solubility
370 of the lipid in the solvent; tributyrin is more water soluble than tricaprylin [41]. This is in accordance
371 with previous studies that demonstrated a higher specificity constant of CALB [40, 42], and other
372 lipase [43] for a short chain substrates compared to long chain counterparts.

373
374 The product information states that the method of lipase immobilization is by physical adsorption to
375 a macroporous polyacrylic resin, but does not give specific information about their immobilization
376 process. The adsorption technique has weaker linkages between the enzyme and support than
377 expected for covalent linkage [11, 33], thus it was of importance to determine whether desorption
378 of lipase could occur. It has previously been demonstrated that physical desorption or leaching of
379 CALB can occur in the presence of detergent such as Triton-X and organic solvents, leading to the
380 conclusion that CALB is most likely physisorbed onto the support by hydrophobic interactions [10,
381 44]. Absence of leaching into the medium was confirmed due to correlation of fatty acid by HPLC
382 and pH stat titration during lipolysis, indicating that the removal of immobilised lipase halts
383 digestion. Consequently, unlike pancreatic lipase, inhibition with a molecular inhibitor such as 4-
384 BPBA is not required [5].

385
386 **Immobilized lipase provides reduced background scattering during *in vitro* lipolysis**
387 SAXS measurements indicated a clear reduction in parasitic scattering in a dispersion of immobilised
388 lipase compared to porcine pancreatin extract, which contains other digestive enzymes such as
389 amylase, trypsin, ribonuclease, deoxyribonuclease, gelatinase and elastase. Interestingly, in the case
390 of the immobilized lipase sample, an upturn was observed at low q . This indicates the presence of
391 some larger particles in the scattering volume which may originate from fragments of the polymer
392 particles.

393

394 The SAXS results following *in vitro* lipolysis of Captex 355 and tricaprylin demonstrated that the
395 structural change that occurs at the “kink” in the sigmoidal profile previously attributed to vesicle
396 formation is retained regardless of lipase used [5]. The decreased parasitic scattering observed
397 during lipolysis with immobilized lipase renders it a better lipase for structural studies. It is not
398 straightforward to simply subtract the buffer and lipase curve from the overall scattering data, as
399 the composition of the bulk phase changes due to the formation of new interfaces and adsorption of
400 these materials during the digestion. Application to time-resolved structural determination through
401 improved quality of scattering data is expected to now be enabled by this discovery.

402

403 **Conclusion**

404 This study provides insight into the use of immobilised lipase for *in vitro* lipolysis experiments. The
405 digestion kinetics of the immobilised lipase depended on the chain length of the substrate, lipid
406 loading and temperature. The structural and compositional changes during *in vitro* lipolysis were
407 maintained regardless of lipase used, however the decreased background scattering observed with
408 immobilised lipase renders it advantageous over lipase sourced from porcine pancreatic extract for
409 understanding structure formation during lipolysis.

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